Atomic Processes in Helium-Krypton and Helium-Xenon Mixtures*

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The momentum transfer collision frequency of thermal electrons with neutrals in a decaying plasma established in helium-krypton and helium-xenon mixtures of known proportions was measured by microwave interferometer at gas temperatures of \sim 200 to 600°K. The energy dependences of the momentum transfer cross sections of electrons with krypton and xenon atoms deduced from these measurements are best repre s ented by $Q_m(u) = 6.56 \times 10^{-15} - 2.79 \times 10^{-14} u^{1/2} + 3.14 \times 10^{-14} u$ and $1.91 \times 10^{-14} - 8.30 \times 10^{-14} u^{1/2} + 9.40 \times 10^{-14} u$ cm², respectively. Here u is the electron energy in electron volts. Mobilities of Kr⁺ and Xe⁺ in helium and their respective parent gas have also been determined, from the characteristic time constants of the electron density decay measured in the afterglow in the mixtures at low pressures, to be $\mu(Kr^+$ in He) = 20.2 \pm 1.2 cm²/V-sec, μ (Kr⁺ in Kr) = 1.01 \pm 0.06, μ (Xe⁺ in He) = 18 \pm 1.1, and μ (Xe⁺ in Xe) = 0.55 \pm 0.03 at ~300°K. A study of the pressure dependence of the characteristic time constants of the electron density decay at fixed ratios of krypton to helium and xenon to helium concentrations yields the three-body conversion frequency of atomic krypton and xenon ions to their respective molecular ions.

I. INTRODUCTION

THE employment of microwave technology in
studying the fundamental atomic collision pro-
cesses in a weakly ionized gas is well known.¹ Neverthe-HE employment of microwave technology in studying the fundamental atomic collision proless, questions have been occasionally raised as to the assumption of thermal equilibrium of the electron gas with the neutrals at times in the afterglow the experiment was performed. In some cases, evidences² showed that the electron temperature could sustain at a level above that of the neutrals at times several hundred microseconds to a few milliseconds after removal of the excitation source. Since almost all physical parameters determined by microwave methods are related directly or indirectly to the electron temperature, it would be appropriate that the electron temperature is measured experimentally. In the present communication, the complete thermalization of the electrons with the neutrals in the He-Kr and He-Xe mixtures is demonstrated by a comparison of the microwave noise emitted from the plasma with that of a standard noise source as detected by a ruby maser. One of the reasons for mixing krypton and xenon with helium is to utilize the helium as a "recoil" gas² for the electrons. Quantitative and qualitative descriptions of various collisional processes are then obtained from the measurements made in the afterglow established in such mixtures. The problems of interest are: (1) The energy dependence of the momentum transfer cross sections of electrons with krypton

and xenon atoms at energies below Ramsauer minimum, (2) the mobilities of thermal Kr^+ and Xe^+ ions in helium and in their respective parent gas at room temperature $(i.e., ~300^{\circ}K), (3)$ the three-body conversion frequency ν_{conv} of atomic krypton and xenon ions to molecular ions³ [see Eq. (7)], and (4) some qualitative evidence in supporting a suggested process of molecular krypton ions formation through collisions of high-lying, shortlived excited atoms with ground state atoms [see Eqs. (14) and (15)], by Hornbeck and Molnar⁴ (HM).

II. EXPERIMENTAL APPARATUS

The gas-handling system is of standard high-vacuum type⁵ baked at \sim 400°C for more than 24 h prior to each sequence of experiments. An ultimate vacuum of the order of 2 to 6×10^{-10} mm Hg is attained. Gases are then introduced into the discharge tube, and the pressure is measured by a (capacity) null reading manometer.⁶ The gases used are mass spectrometer controlled grade supplied by Linde Air Products Company. The discharge tube is made of thin wall (0.7 mm thick) Pyrex tubing of 22 mm outside diameter and 72 cm long with 6 cm tapering to a point at each end. The tube is housed coaxially in a 1-in. \times 1-in. square waveguide which is connected to the standard x -band waveguide system through two 6-in. tapering sections. The gas is ionized by a variable high voltage dc pulse of several thousand volts and seven microseconds duration repeated at a frequency of 31.2 cps. The electrodes of the discharge tube are made out of high-purity titanium for its good gettering property for the atmospheric gases.⁷

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¹The pertinent references can be found in the review articles of L. Goldstein, *Advance in Electronics and Electron Physics* (Academic Press Inc., New York, 1955) Vol. VIII, p. 473; S. C. Brown, *Basic Data of Plasma Physics* (Tech Press, Cambridge, Massa-chusetts, and John Wiley & Sons

national Conference on Ionization Phenomena in Gases, Uppsala, 1959, edited by N. R. Nillson (North-Holland Publishing Company, Amsterdam, 1960), Vol. I, p. 99; M. A. Biondi, Phys. Rev. 90, 730 (1953).

³This is similar to the three-body He₂⁺ formation process studied by A. V. Phelps and S. C. Brown, Phys. Rev. 86, 102 (1952).

⁴ J. A. Hornbeck and J. P. Molnar, Phys. Rev. 84, 621 (1951). 6 D. Alpert, J. Appl. Phys. 24, 860 (1953).

⁶ The design is originated from Westinghouse Research Labora-

tories, West Pittsburgh, Pennsylvania. A unique package is now marketed by Granville-Phillips Company, Boulder,|Colorado. ⁷V. L. Stout and M. D. Gibbons, J. Appl. Phys. 26, 1488 (1955).

Electron density variations and the effective electron collision frequency v_{eff} for momentum transfer are measured by microwave interferometry in the decaying plasma created in helium-krypton and helium-xenon mixtures. A schematic diagram of the microwave circuitry used in part of the experiment is shown in Fig. 1. A low-power (\sim 2 μ W), 9.03 or 8.53 kMc/sec probing signal (continuous or pulsed) is employed to measure the phase shifts and attenuations on the microwave due to the presence of plasma. The temperature of the discharge tube is monitored constantly by three copperconstantan thermocouples. The ruby maser is operated at a pump frequency of \sim 21kMc/sec and a signal frequency of 8.53 kMc/sec with a gain of \sim 25 dB. A typical result is shown in Fig. 2.

FIG. 1. Schematic diagram of one of the microwave circuitries employed in the present experiment.

III. MOMENTUM TRANSFER COLLISION CROSS SECTION

It has been shown⁸ that the effective electron collision frequency for momentum transfer v_{eff} , under the influence of low-level dc or rf electric field, in a weakly ionized gas, is given by

$$
\nu_{\rm eff} = \nu_{\rm em} + \nu_{\rm ei},\tag{1}
$$

where ν_{em} and ν_{ei}^9 are the momentum transfer collision frequencies of electrons with neutrals and ions, re-

FIG. 2. A direct comparison of the noise emitted by the decaying plasma, created in He-Xe mixture of 54.6% Xe and a total gas pressure of 4.83 mm Hg, with that from a standard noise source of 300° K. The absorptivity of the plasma is checked with $a\sim 2 \mu W$ 8.53 kMc/sec coherent radiation and is found to be 1 up to 450 yusec in the afterglow. The photograph shows that the electrons relax back to the gas temperature approximately 300μ sec after termination of the pulse.

spectively. Furthermore.⁸

$$
v_{em} = \frac{N_m}{3} \left(\frac{2}{\pi}\right)^{1/2} \left(\frac{m}{kT_e}\right)^{5/2} \int_0^{\infty} Q_m(v) v^5 \exp\left(-\frac{mv^2}{2kT_e}\right) dv. \tag{2}
$$

Here N_m is the number densities of the neutrals, m is the electron mass, v the electron velocity, and T_e the electron temperature, *k* is the Boltzmann's constant and $Q_m(v)$ is the momentum transfer cross section of electrons with the neutrals.

For binary mixtures as in the present experiment, $\nu_{em} = \nu_{em} + \nu_{em}$, and

$$
\frac{\nu_{em}}{p_t} = \left(\frac{\nu_{em_1}}{p_1} - \frac{\nu_{em_2}}{p_2}\right) \frac{p_1}{p_t} + \frac{\nu_{em_2}}{p_2},\tag{3}
$$

*P_t p*₂ *p*₂ *p*₂ *p*₂ *p*₂ *p*₂ *p*₂ *p*₂ *p*₂ *p*₂ where p_t is the total gas pressure and is equal to the sum

FIG. 3. ν_{em}/p_t versus percentage of helium in helium-krypton and helium-xenon mixtures at 200 and 303°K. The straight-line behavior is predicted by Eq. (3) in the text under the condition of $(\nu/\omega)^2 \ll 1$. The two ordinates of ν_{em}/p_t at 0 and 100% He give $\nu_{em}(\text{Kr})/p(\text{Kr})$, $\nu_{em}(\text{Xe})/p(\text{Xe})$, and $\nu_{em}(\text{He})/p(\text{He})$ at the tem-
peratures indicated.

⁸ V. L. Ginsburg and A. V. Gurevich, Usp. Fiz. Nauk 70, 201 (1960) [translation: Soviet Phys.—Usp. 3, 115 (1960)]; V. N.
Kolesnikov and V. V. Obukhov-Denisov, Zh. Eksperim. i Teor.
Fiz. 42, 1901 (1962) [translation: Soviet Phys.—JETP, 15, 692
(1962)]; C. L. Chen and M. Raether, Phy Equation (1) is correct only when $(\omega/\nu)^2 \gg 1$ where ω is the radian frequency of the applied electric field and ν is the momentum transfer collision frequency. In our experiment, $(\omega/\nu)^2 \approx 100$.
 A brief summary

^{1037 (1955).} The formula for ν_{ei} adopted in the present article is
that by V. L. Ginsburg, J. Phys. U.S.S.R. 8, 253 (1944) and has
been shown approximately correct experimentally at 300°K by
Anderson and Goldstein. Th tation of the present data is minute since $\nu_{em} = \nu_{eff} - \nu_{ei}$ is always checked in the early and the very late afterglow where ν_{ei} is presumably negligible.

FIG. 4. $\nu_{em}(Kr)/\rho(Kr)$ versus T_e . The various symbols on the graph represent the values of $\nu_{em}(Kr)/p(Kr)$ deduced from dif-
ferent fractional krypton concentrations: \bullet from Fig. 3, *41.6%
Kr, \bigcirc 50.1% Kr, $+$ 58.8% Kr, \bigcirc 73% Kr. The solid curve is the
best fit to the exper suming $Q_m(v) = A + Bv + Cv^2$

of the partial pressures p_1 and p_2 of the mixtures. All pressures are hereafter referred to 0°C. Therefore, the ratio of measured electron-molecule collision frequency to the total gas pressure is a linear function of fractional concentration of one of the species. In our experiments, ν_{ei}/ν_{eff} is of the order of 1 to 10% in the afterglow in which ν_{eff} is measured and⁹ ν_{ei} is subtracted out from ν_{eff} to get ν_{em} . The results are shown in Fig. 3. From the extrapolated values of ν_{em}/p_t at $p(\text{He})/p_t = 1$ and 0, the momentum transfer collision probability¹ \overline{P}_m of electrons with He, Kr, and Xe atoms are found to be: 18.9, 54.7, and 151 cm^2/cm^3 , respectively, at 303°K and 18.9, 77.5, and 221 cm²/cm³, respectively, at 200 $\rm{^\circ K}$. Thus, within experimental accuracy, \vec{P}_m (He) and hence $Q_m(v)$ for He is a constant in this range. This fact agrees with what has been found by other investigators¹¹ and is utilized later in deducing $\nu_{em}(Kr)/\nu(Kr)$ and $\nu_{em}(Xe)/$ $p(Xe)$ from ν_{eff}/ρ_t measured at higher temperatures. The temperature dependence of $\nu_{em}(\text{He})/p(\text{He})$ is taken to be

$$
\nu_{em}(\text{He})/p(\text{He}) = 1.56 \times 10^{7} T_{e}^{1/2} \text{ sec}^{-1} \text{mm Hg}^{-1}. \quad (4)
$$

By subtracting electron-ion⁹ and electron-helium [as calculated from Eq. (4)] contributions from ν_{eff} , the resulting momentum transfer collision frequency of electrons with Kr and Xe atoms as a function of electron temperature is shown in Figs. 4 and 5. The velocity dependence of the momentum transfer cross section $Q_m(v)$ is determined from a best fit to the experimental

points by assuming $Q_m(v) = A + Bv + Cv^2$ cm² in Eq. (2). The solid curves on Figs. 4 and 5 are the best fits. The derived energy dependence of the cross sections are

$$
Q_m(u) = 6.56 \times 10^{-15} - 2.79 \times 10^{-14} u^{1/2}
$$

for krypton and

$$
Q_m(u)\!=\!1.91\!\times\!10^{-14}\!-\!8.30\!\times\!10^{-14}u^{1/2}
$$

$$
+9.40\times10^{-14}u
$$
 cm² (6)

(5)

 $+3.14\times10^{-14}$ u cm²

for xenon. Here *u* is the electron energy in electron volts.

Recently, Pack, Volshall, and Phelps (PVP)¹² have deduced $Q_m(u)$ from their electron mobility studies in Kr and Xe. Their results together with the present one are shown in Figs. 6 and 7, in which PVP's notation are preserved. O'Malley¹³ has adopted "atomic effective range formulas"¹⁴ to analyze Ramsauer-Kollath (RK) scattering experiments.¹⁵ In this analysis, the parameters of the theory are so chosen to fit RK experimental cross sections. These calculations were extrapolated to zero energy and are also shown in Figs. 6 and 7. All agree fairly well with each other in shape but not in absolute value. The disagreements can be atrributed partly to the approximations and experimental errors in each case.

IV. ION MOBILITIES AND CONVERSION FREQUENCIES

The main electron loss process in the afterglow of a low pressure, weakly ionized noble gaseous discharge is ambipolar diffusion.¹⁶ For the present experiments

Fig. 5. $\nu_{em}(\text{Xe})/p(\text{Xe})$ versus T_e . The various symbols on the graph represent the values of $\nu_{em}(\text{Xe})/p(\text{Xe})$ deduced from different fractional xenon concentrations: \bullet from Fig. 3, +50% Xe, \circ 58.8% Xe, curve is the best fit to the experimental points according to Eq. (2) and assuming $Q_m(v) = A + Bv + Cv^2$.

12 J. L. Pack, R. E. Voshall, and A. V. Phelps, Phys. Rev. **127,** 798 (1962).

- 13 T. F. O'Malley, Phys. Rev. **130,** 1020 (1963). 14 T. F. O'Malley, L. Spruch, and L. Rosenberg, J. Math. Phys. 2, 491 (1961); Phys. Rev. **125,** 1300 (1962).
- 15 C. Ramsauer and R. Kollath, Ann. Physik **12,** 837 (1932).

16 M. A. Biondi and S. C. Brown, Phys. Rev. **75,** 1700 (1949); W. P. Allis and D. J. Rose, *ibid.* 93, 84 (1954); I. B. Bernstein and T. Holstein, *ibid.* 94, 1475 (1954).

¹⁰ It is easily shown from the rf electrical conductivity that $\nu_{eff} = \frac{4}{3} N_m \bar{Q}_m \langle v \rangle$, where \bar{Q}_m is the effective momentum transfer cross section and $\langle v \rangle = (8kT_e/\pi m)^{1/2}$. Then, the effective momentum trans Laboratory, $\begin{bmatrix} \text{unpublished} \\ \text{unpublished} \end{bmatrix}$.

[&]quot; L . Gould and S. C. Brown, Phys. Rev. 95, 897 (1954); J. L. Pack and A. V. Phelps, *ibid.* **121,** 798 (1961).

(helium-krypton or helium-xenon mixtures of various proportions) the ions created in the active discharge are believed to be atomic krypton or xenon ions when suitable breakdown voltage pulse is employed.¹⁷ This is supported by the spectral examination of the discharge in He-Kr mixtures with a Bausch and Lomb Littrow No. 5402 Spectrograph which has a dispersion of 7.7 \AA /mm at 3670 Å and 45.5 Å/mm at 6700 Å. It is found that no band spectra of any kind and only atomic krypton lines are presented. HM noticed in their mass spectrometric studies of molecular ions in noble gases that Kr_2 ⁺ and Xe_2 ⁺ ions are much more difficult to be formed than He₂⁺, Ne₂⁺, and Ar₂⁺ through collisions of highlying, short-lived excited states with the ground state atoms. Therefore, at low gas pressures and careful b reakdown condition,¹⁷ the molecular ions formed by this process can be ignored. However, Kr^+ or Xe^+ ions created in the active discharge, while diffusing through the mixture to the walls in a decaying plasma, experience not only elastical scatterings from helium and their parent gas atom, but also may change their identities to molecular ions through three-body collisions:

$$
{\begin{cases}\nKr^+ \\
Xe^+\n\end{cases}} + {\begin{cases}\nKr \\
Xe\end{cases}} + He \rightarrow {\begin{cases}\nKr_2^+ \\
Xe_2^+\n\end{cases}} + He. \tag{7}
$$

$$
n(t) = A \exp(-t/\tau_A) + B \exp(-t/\tau_M), \qquad (8)
$$

where

$$
1/\tau_A = D_{a_A}/\Lambda^2 + \nu_{\text{conv}} \tag{9}
$$

and

$$
1/\tau_M = D_{aM}/\Lambda^2. \tag{10}
$$

Here *n* is the electron density, Λ is the characteristic diffusion length of the discharge tube, D_{a_A} and D_{a_M} are the ambipolar diffusion coefficients of the atomic and molecular ions in the mixtures, and $\nu_{\text{conv}}=C_{\text{conv}}p_1p_2$ is the atomic to molecular ion conversion frequency. *A* and B are constants related to the initial atomic and molecular ion concentrations and the physical constants just mentioned above. If $\nu_{\rm conv}$ is smaller than the molecular ion diffusion rate, and *B<KA* (which is true in our experiment),¹⁷ the electron loss rate in the afterglow is simply described by a single time constant, i.e., τ_A . From Eq. (9) it is easy to show that the product of ambipolar diffusion coefficient D_a of electrons to the partial pressure of helium in the helium-krypton mixtures, for example, takes the following form:

$$
D_a p(\text{He}) = \frac{T_e}{7.63} \frac{\mu(\text{Kr}^+ \text{ in He})}{1 + \left[\frac{p(\text{Kr})}{p(\text{He})}\right] \left[\frac{\mu(\text{Kr}^+ \text{ in He})}{\mu(\text{Kr}^+ \text{ in He})}{\mu(\text{Kr}^+ \text{ in Kr})}\right]} + \Lambda^2 C_{\text{conv}} p^2(\text{He}) p(\text{Kr}) \,,\tag{11}
$$

where μ (Kr⁺ in He) and μ (Kr⁺ in Kr) are the mobilities of atomic krypton ions in helium and in krypton, respectively, referred to 0^oC and 760 mm Hg gas pressure. $p(Kr)$ and $p(He)$ are the partial pressures of krypton and helium in the mixtures. In deriving Eq. (11), Einstein's relation¹⁹ and Blanc's law²⁰ have been employed. At very low pressures, the second term to the right-hand side of Eq. (11) can be neglected, and

$$
D_a \phi(\text{He}) = \frac{T_e}{7.63} \frac{\mu(\text{Kr}^+ \text{ in He})}{1 + \left[\frac{p(\text{Kr})}{p(\text{He})}\right] \left[\frac{p(\text{Kr}^+ \text{ in He})}{p(\text{Kr}^+ \text{ in He})}{\mu(\text{Kr}^+ \text{ in He})}\right]}.
$$
(11a)

 μ (Kr⁺ in He) and μ (Kr⁺ in Kr) are then determined from the best fit of the measured quantities $D_a p(\text{He})$ and the percentage of Kr (or of Xe in the case of He-Xe mixtures) at a constant temperature according to Eq. (11a). These are shown in Fig. 8. The best fit yields μ (Kr⁺ in He) = 20.2 \pm 1.2 cm²/V-sec, μ (Kr⁺ in Kr) $= 1.01\pm0.06$, μ (Xe⁺ in He) = 18 \pm 1.1, and μ (Xe⁺ in Xe) $= 0.55 \pm 0.03$ at 303°K.

The theoretical calculated mobilities of thermal energy ion relevant to the present experiment, together with the values determined by other authors, are presented in Table I. The theoretical values of $\mu ($ Kr⁺ in ${\rm He})$

and μ (Xe⁺ in He) are calculated by the use of Langevin's theory in the polarization limit.²¹ The dielectric constant for helium adopted here is that recommended by Maryott and Buckley.²²

As the total pressure increases while keeping the percentage of krypton in the mixtures fixed, the measured values of $D_a p(\text{He})$ should vary linearly with

¹⁷ The breakdown voltage pulses are adjusted in strength (keeping the duration fixed at $\sim 7\mu$ sec and repeated at 31.2 cps) just enough to maintain a stable, repeatable, pulsed discharge. In this manner, the excitation light is usually very dim. Therefore, the initial concentration of molecular ions formed by HM process is believed to be negligibly small. See Ref. 4 and Sec. IV of the text.

¹⁸ Electron decay through electron-ion recombination processes has been neglected in the present treatment for its ineffectiveness under the present experimental conditions. The theoretical treatments of this subject is summarized in the book: *Atomic and Molecular Processes,* edited by D. R. Bates (Academic Press Inc.,

New York, 1962).

¹⁹ W. P. Allis, in *Handbuch der Physik*, edited by S. Flügge (Springer-Verlag, Berlin, 1956), Vol. XXI.

²⁰ L. B. Loeb, *Basic Processes in Gaseous Electronics* (University

of California Press, Ber

FIG. 6. Momentum transfer cross section of electrons with krypton atoms. The result of the present experiment is compared with those found by PVP and the theoretical calculations by O'Malley.

TABLE I. Comparison of experimental and theoretical values of Kr^+ and Xe^+ mobilities (in cm²/V-sec).

\smallsetminus Ion Gas ⁸	Kr^{+} Experiment Theory		$Xe+$ Experiment Theory	
He Кr	$20.2 + 1.2$ * $0.9 - 0.95b$ 0.90 ^c $0.01 + 0.06$ ^a	17.0 1.0 ^d 0.9 ^e	$18 + 1.1^a$ \cdots	16.8
Xe	\cdots		$0.6 - 0.65b$ 0.58 ^c $0.55 + 0.03$ *	0.66d 0.60 ^e

b R. N . Varney, Phys. Rev. **88,** 362 (1952). ° M. A. Biondi and L. M. Chanin, Phys. Rev. 94, 910 (1954). d I. B. Bernstein (unpublished). • A. Dalgarno, Phil. Trans. Roy. Soc. London **A250,** 426 (1958).

 p^2 (He) p (Kr) should there be the three-body molecular ion formation process. The slope of $D_a\phi$ (He) versus $p^2(\text{He})p(\text{Kr})$ yields the value of $C_{\text{conv}}\Lambda^2$, and should be

FIG. 7. Momentum transfer cross section of electrons with xenon atoms. The result of the present experiment is compared with those found by PVP and the theoretical calculations by O'Malley.

independent of the krypton percentage. Figure 9 presents the results in He-Kr mixtures for four different krypton percentages; i.e., 1.7, 4.3, 5.97, and 14% . The slopes are fairly well the same, and *Cconv* so determined is (76 ± 4) mm Hg⁻²-sec⁻¹, i.e.,

$$
\nu_{\rm conv} = (76 \pm 4) p(\text{He}) p(\text{Kr}). \tag{12}
$$

Similar studies are also made for Xe⁺ to Xe₂⁺ conversion. Typical results are shown in Fig. 10. In this case, $C_{\text{conv}}^{\dagger} = (140 \pm 9)$ mm Hg⁻²-sec⁻¹, and

$$
\nu_{\text{conv}} = (140 \pm 9) p(\text{He}) p(\text{Xe}). \tag{13}
$$

Similar to the gas-kinetic conditions²⁸ in two-body charge or excitation transfer collisions, it is reasonable to believe that the lesser the amount of energy carried away by the third body in reaction (7), the higher the probability of molecular ion formation. Then the larger

FIG. 8. *Dap* (He) versus percentage of krypton or xenon in helium-krypton or helium-xenon mixtures. The solid curves are the best fits according to the functional form of Eq. (11a).
From these, it is determined that μ (Kr⁺ in He)=20.2 \pm 1.2 cm²/V-sec, μ (Kr⁺ in Kr)=1.01 \pm 0.06, μ (Xe⁺ in He)=18 \pm 1.1, and μ (Xe⁺ in Xe

value of C_{conv} , which is proportional to the formation probability, for Xe^{+} to Xe^{+} than Kr^{+} to Kr_{2}^{+} indicates that the amount of energy carried away by He is smaller in the former than in the latter case. Should this be so, the binding energy of Xe_2 ⁺ would be smaller than that of $Kr₂$ ⁺. This has to await a further study of the appearance potentials in these gases to confirm it. Nevertheless, observations by $HM⁴$ seemed to suggest the same explanation. They noticed, in their mass-spectrometry studies of molecular ions formed by electron bombardment in noble gases, that the current peaks of $Xe⁺$ to Xe_2 ⁺ is 4×10^4 to 1 while Kr⁺ to Kr₂⁺ is 2×10^4 to 1. The apparent greater difficulty in Xe_2 ⁺ formation than Kr_2 ⁺ through (taking xenon as an example)

$$
e + \text{Xe} \to \text{Xe}^* + e, \tag{14}
$$

$$
Xe^* + Xe \to Xe_2^+ + e, \tag{15}
$$

23 H. S. W. Massey and E. H. S. Burhop, *Electronic and Ionic Impact Phenomena* (Clarendon Press, Oxford, England, 1952).

could be explained as that Xe* (stands for xenon atom in a high-lying, short-lived excited state) required for the reaction must be very close to the ionization limit if the binding energy of $\check{\rm Xe_2^+}$ were very small. The excitation cross section is known to drop off rapidly, in general, as the total quantum number increases.²⁸ Therefore, their finding seems to be in harmony with C_{conv} determined here.

We have also studied qualitatively the molecular ion formation processes proposed by HM [see Eqs. (14)] and (15)]. We observe that the characteristic time constant of the electron density decay is a strong function of the excitation light in the active discharge, while keeping p_t and the relative concentration of Kr (or Xe) unchanged. Qualitatively, the brighter the excitation light, the smaller the characteristic ambipolar diffusion time constant. Since electrons have already relaxed back to the gas temperature at times in the afterglow the measurements were made and high-order modes of diffusion are believed not to exist at such late times (2 to 12 msec) in the afterglow. The only feasible explanation offered to such pheonmenon is the formation of molecular ions through processes (14) and (15). Xe_2 ⁺ or Kr_2 ⁺

FIG. 9. $D_a p$ (He) versus p^2 (He) p (Kr). The slope is proportional to the conversion frequency of reaction (7).

FIG. 10. $D_a p$ (He) versus p^2 (He) p (Xe). The slope is proportional to the conversion frequency of reaction (7).

ions are known to have a higher mobility than Xe⁺ or Kr⁺ in their parent gases due principally to lack of charge exchange.²⁴ The light intensity in the active discharge is interpreted as an indirect measure of Xe* or Kr* concentrations. No detailed correlations between the distribution of line intensities and the molecular ion concentrations are pursued at the present time. Further mass- and optical-spectrometry studies are necessary.

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24 R. N. Varney, Phys. Rev. 88, 362 (1952); M. A. Biondi and L. M. Chanin, *ibid.* 94, 910 (1954).

Fig. 2. A direct comparison of the noise emitted by the decaying
plasma, created in He-Xe mixture of 54.6% Xe and a total gas
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